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Synthesis of *C*2-functionalized pyrimidines from 3,4-dihydropyrimidin-2(1*H*)-ones by the Mitsunobu coupling reaction

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ABSTRACT

The Biginelli 3,4-dihydropyrimidin-2(1*H*)-one was converted to various *C*2-multifunctionalized pyrimidines via the dehydrogenation and Mitsunobu reaction using amines, alcohols, phenols and carboxylic acids as nucleophiles. A possible mechanism was also proposed to rationalize the formation of products.

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1. Introduction

Biginelli 3,4-dihydropyrimidin-2(1*H*)-ones (DHPMs)¹ are central subunits in a broad range of medicinal agents, which display interesting pharmacological and biological properties, such as calcium channel modulators, α_{1a} -adrenergic receptor antagonists, mitotic kinesin inhibitors and hepatitis B virus replication inhibitors.² The DHPM core was also found in several marine derived natural products, such as Crambine, Batzelladine B (potent HIV gp-120CD4 inhibitors) and Ptilomycalin alkaloids.³ Additionally, the Biginelli DHPMs are important building blocks in synthesis of multifunctionalized pyrimidines. Although much effort has been paid on the development of methods for the synthesis of the pyrimidine derivatives,⁴ few methodologies are available towards efficiently synthesis of 2-substituted pyrimidines. In general, C2-substituted pyrimidines were obtained from Biginelli DHPMs by a four-step strategy involving sequentially dehydrogenation, tautomerization, activation and coupling with a nucleophile^{5,6} (Scheme 1). However, the harsh condition in chlorination using POCl₃ at high temperatures would threaten those substrates with sensitive functional groups (Scheme 1, Eq. 1).⁵ Besides, S-alkylation was absolutely necessary for sulfide oxidation (Scheme 1, Eq. 2). 6b,e

Dehydrogenation of the Biginelli DHPMs is also a well known difficult process leading to low yield, when oxidizing agents, such as SeO₂, DDQ or HNO₃ were used.⁷ It was not until 2005 that a more efficient approach was achieved by Kang et al., in which⁸ Biginelli DHPMs were converted to the *C*2-functionalized pyrimidines via Kappe dehydrogenation^{7c} and PyBroP-mediated coupling with nucleophiles at rt for 24 h (Scheme 1, Eq. 3). At the same time, Yamamoto et al.⁹ developed a mild and practical procedure for dehydrogenation of DHPMs using *tert*-butylhydroperoxide (TBHP), which can be applied on a large scale.

However, some limitations still exist in the synthesis of C2functionalized pyrimidines based on the methodologies mentioned above, such as elevating temperature, using of expensive catalyst, needing long reaction time or multistep. Thus, it is necessary to develop simpler and more efficient methods under mild reaction conditions to construct these compounds. To address this problem. we have developed a mild and rapid procedure to the synthesis of C2-substituted pyrimidines by sequential functionalization of 3,4dihydropyrimidine-2(1H)-ones via oxidation and esterification followed by cross-coupling reaction with N, S and O nucleophiles at rt (Scheme 1, Eq. 4).¹⁰ The results showed that the method is not essential for the reaction of pyrimidin-2-yl sulfonates with phenols to provide phenolic pyrimidine derivative. Herein we report a general and comprehensive strategy for the preparation of highly C2-functionalized pyrimidines (4, 6, 8) including phenolic pyrimidines through direct functionalization of the 2-hydroxy pyrimidine

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Scheme 1. Synthesis of the C-2 functionalized pyrimidines deriving from DHPMs.

core **2**, which obtained from Biginelli **3**,4-dihydropyrimidine-2 (1*H*)-ones **1** (Scheme 2). The OH group of the 2-hydroxy pyrimidine can be replaced by different nitrogen, oxygen and acid nucleophiles using amines **3**, alcohols (or phenols) **5** and benzoic acids **7** via Mitsunobu reaction.

Mitsunobu reaction. The results were summarized in Table 2. No matter what substituent groups were on the aromatic rings of 2-hydroxy pyrimidine, the reactions proceeded smoothly, and the cross-coupling products were isolated in good yields. Cyclic secondary amines were the most successful in this process (entries

$$R_{2} \xrightarrow{NH} \xrightarrow{TBHP} R_{2} \xrightarrow{N} \xrightarrow{NuH 3, 5, 7} R_{2} \xrightarrow{NuH 3, 5, 7} R_{2} \xrightarrow{NuH 3, 5, 7} R_{1} \xrightarrow{NuH 3, 5, 7} R_{2} \xrightarrow{NuH 3, 5} R_{2}$$

Scheme 2. Preparation of C2-multifunctionalized pyrimidines.

2. Results and discussion

The Biginelli DHPM 1 and their dehydrogenated compounds 2 were readily prepared according to the procedures reported by Fu et al.¹¹ and Yamamoto et al.⁹ As a starting point in our investigations, we examined the Mitsunobu reaction between model substrate 2-hydroxy pyrimidine 2d and piperidine 3a. Firstly, we chose diethyl azodicarboxylate (DEAD) and triphenylphosphine (TPP) as Mitsunobu regents. ¹² As expected, when the reaction was performed at rt for 12 h, the cross-coupling product 4dA was obtained in low yield together with the isolation of a C2 ethoxy substituted pyrimidine 4dB (Table 1, entries 1 and 2). When the reaction was performed at 0 °C to rt, the yield of **4dA** was increased obviously (entry 3). Diisopropyl azodicarboxylate (DIAD) exhibited the best activity and higher yields of 4dA were obtained with shorter reaction time (entries 7 and 8). Survey of various conditions indicates that the optimum ratio of 2d:3a:DIAD:Ph₃P is 1:1.5:1.5:1.5.

Next, the addition order of the reactants and solvents was evaluated. Initially, the reaction was conducted under the typical procedure of the Mitsunobu reaction as follows: DIAD was added dropwise to the mixture of TPP, **2d**, and **3a**, but the reaction did not occur smoothly and only 56% yield of **4dA** was achieved (entry 9). A survey of the adding order showed that the yield could be improved significantly according to the following order: at 0 °C, DIAD (1.5 mmol) was dissolved in THF (3 mL), followed by the addition of TPP with the formation of a suspension, then **2d** and **3a** were added (entry 8). The mixture was warmed to rt, and stirred for further 2.5 h to give **4dA** in 78% yield.

With the optimized conditions in hand, we then examined the scope of 2-hydroxy pyrimidine substrates by varying the substituents on the aromatic ring to explore the generality of this

Table 1Optimization of reaction conditions for product **4dA**^a

Entry	Equiv of 2d:3a	Reagent	Temp (°C)	Time (h)	Isolated yield (%)	
					4dA	4dB
1	1:1	DEAD	25	12	10	60
2	1:1.5	DEAD	25	12	11	63
3	1:1.5	DEAD	0-25	12	50	40
4	1:1.5	DEAD	60	12	10	68
5	1:1.5	DEAD	0-25	2.5	56	35
6^{b}	1:1.5	DEAD	0-25	2.5	53	39
7	1:1	DIAD	0-25	12	56	_
8	1:1.5	DIAD	0-25	2.5	78	_
9 ^c	1:1.5	DIAD	0-25	2.5	56	_

^a Reaction conditions: at 0 °C, DIAD (1.5 mmol) was first dissolved in THF (3 mL), followed by the addition of TPP (1.5 mmol) with the forming of a suspension, then 2d (1.0 mmol) and 3a (1.5 mmol) were added. The temperature was increased to rt.

b CH₂Cl₂ was used as solvent.
 c Reaction conditions: DIAD (1.5 mmol) was added dropwise to the mixture of TPP (1.5 mmol), 3a (1.5 mmol) and 2d (1.0 mmol).

1–14). Acyclic amines also gave promising results (completed within 5 h), albeit a bit longer time was needed (entries 15). However, when aromatic amine was involved, the reaction became complicated, and most of the starting 2-hydroxy pyrimidine **2** was recovered (not shown in Table 2).

Table 2
Coupling of 2-hydroxy pyrimidines 2 with NH nucleophiles 3^a

Entry	Ar/ 2	R_1	R ₂	NH	4	Yield (%) ^b
1	Ph/ 2a	Me	OEt	HN 3a	4a	84
2	4-MeO-Ph/2b	Me	OEt	3a	4b	81
3	4-Me-Ph/ 2c	Me	OEt	3a	4c	83
4	4-Cl-Ph/ 2d	Me	OEt	3a	4d	78
5	4-Br-Ph/ 2e	Me	OEt	3a	4e	85
6	4-F-Ph/2f	Me	OEt	3a	4 f	86
7	4-F-Ph/ 2g	i-Pr	OMe	3a	4g	80
8	Ph/ 2a	Me	OEt	HN 3b	4h	82
9	4-MeO-Ph/2b	Me	OEt	3b	4i	80
10	4-Me-Ph/ 2c	Me	OEt	3b	4j	80
11	4-Cl-Ph/ 2d	Me	OEt	3b	4k	83
12	4-Br-Ph/ 2e	Me	OEt	3b	41	80
13	4-F-Ph/ 2f	Me	OEt	3b	4m	82
14	4-F-Ph/2g	i-Pr	OMe	3b	4n	84
15 ^c	Ph/ 2a	Me	OEt	CH ₃ CH ₂ NH ₂ 3c	40	77

 $^{^{\}rm a}$ Reaction conditions: at 0 °C, DIAD (1.5 mmol) was first dissolved in THF (3 mL), followed by the addition of TPP (1.5 mmol) with the forming of a suspension, then **2** (1.0 mmol) and **3** (1.5 mmol) was added. The temperature was increased to rt and stirred for another 2.5 h.

- b Isolated yields.
- ^c Reaction was completed within 6 h.

When this procedure was applied to the reaction between 2-hydroxy pyrimidine 2 and alcohols 5, to our delight, the target molecules were obtained in excellent isolated yields (Table 3). As indicated in Table 3, the reaction could proceed smoothly (entries 1–4). Then alcohol was changed to phenols, and the reaction could also occur smoothly with phenols bearing electron-withdrawing groups, such as nitro and chloro, affording desired products 6e-i good to excellent isolated yields (entries 5-9) with prolonged reaction time. Unfortunately, product 6j could not be isolated from the reaction mixture by column chromatography and the corresponding yield was determined by ¹H NMR (entry 10). When phenols with an electron-donating group (such as CH₃O) were used, the reaction gave the desired product in trace yield (detected by TLC and ¹H NMR) (entry 11). The results implied that the phenols bearing electron-withdrawing groups were preferred in this reaction, which was in agreement with other Mitsunobu reactions. It is because that the phenols with lower pK_a value can accelerate this cross-coupling reaction. 12a

To further demonstrate the utility of the Mitsunobu reaction, we also tested the reaction between 2-hydroxy pyrimidine **2** and carboxylic acids **7**. As we expected, Mitsunobu reaction occurred smoothly giving *C2* benzoyloxy pyrimidines **8a**–**g** (Table 4). All substrates, either 2-hydroxy pyrimidines or carboxylic acids, with either electron-rich (CH₃O) or electron-deficient (NO₂, Cl) aryl groups, afforded the corresponding *C2* benzoyloxy pyrimidines **8a**–**g** in good to high yields, in favor of the one with electron-deficient groups.

As mentioned above, the nuclephiles, such as phenols and carboxylic acids with lower pK_a prefer the cross-coupling reaction. Thus the mechanism was proposed for the formation of C2-multifunctionalized pyrimidines via the Mitsunobu reaction based on a general pathway as shown in Scheme 3. The first step is the irreversible formation of the Morrison–Brunn–Huisgen (MBH) betaine $\mathbf{9}$. In step 2, this betaine $\mathbf{9}$ deprotonates the nucleophile to form the ionic specie $\mathbf{10}$, which reacted with the 2-hydroxy

Table 3Coupling of 2-hydroxy pyrimidines **2** with OH nucleophiles **5**^a

Entry	Ar	R ₁	R ₂	Time (h)	ROH 5	6	Yield (%) ^b
1	Ph	Me	OEt	2	CH ₃ CH ₂ OH	6a	95
2	Ph	Me	OEt	2	(CH ₃) ₂ CHOH	6b	93
3	Ph	Me	OEt	2	PhCH ₂ OH	6c	90
4	4-Cl-Ph	Me	OEt	2	CH₃CH₂OH	6d	96
5	Ph	Me	OEt	6	$HO NO_2$	6e	82
6	4-MeO-Ph	Me	OEt	6	$HO NO_2$	6f	95
7	4-Cl-Ph	Me	OEt	6	HO-\(\bigs_\)-NO2	6g	86
8	4-F-Ph	i-Pr	OMe	6	HO——NO ₂	6h	80
9	4-Cl-Ph	Me	OEt	6	но-{_}_сі	6i	78
10	4-Cl-Ph	Me	OEt	12	но-	6j	60 ^c
11	4-Cl-Ph	Me	OEt	12	HO—OMe	6k	Trace ^d

 $^{^{\}rm a}$ Reaction conditions: at 0 °C, DIAD (1.5 mmol) was first dissolved in THF (3 mL), followed by the addition of TPP (1.5 mmol) with the forming of a suspension, then $\bf 2$ (1.0 mmol) and $\bf 5$ (1.5 mmol) was added. The temperature was increased to rt and stirred for another 6 h.

- b Isolated yields.
- ^c NMR yield is determined by ¹H NMR spectroscopy.
- ^d Detected by TLC and ¹H NMR.

Table 4Coupling of 2-hydroxy pyrimidines **2** with carboxylic nucleophlies **7**^a

Entry	Ar	R ₁	R ₂	7	8	Yield (%)b
1	Ph	Me	OEt	PhCOOH	8a	70
2	4-MeO-Ph	Me	OEt	PhCOOH	8b	68
3	4-Cl-Ph	Me	OEt	PhCOOH	8c	69
4	4-F-Ph	i-Pr	OMe	PhCOOH	8d	63
5	4-Cl-Ph	Me	OEt	4-MeO-PhCOOH	8e	72
6	4-Cl-Ph	Me	OEt	4-NO ₂ -PhCOOH	8f	80
7	4-Cl-Ph	Me	OEt	4-Cl-PhCOOH	8g	79

^a Reaction conditions: at 0 $^{\circ}$ C, DIAD (1.5 mmol) was first dissolved in THF (3 mL), followed by the addition of TPP (1.5 mmol) with the forming of a suspension, then **2** (1.0 mmol) and **7** (1.5 mmol) was added. The temperature was increased to rt and stirred for another 6 h

pyrimidine **2** providing the key alkoxyphosphonium salt **11** and the hydrazine RO₂CNH—NHCO₂R. After nucleophilic substitution and releasing of a molecule of triphenylphosphine oxide, the product, *C*2-multifunctionalized pyrimidine, is formed.

It is noteworthy that the study reported above, to our best knowledge, is the first general exploration of cross-coupling reaction of 2-hydroxy pyrimidines with N, O and acid nucleophiles to give *C*2-substituted pyrimidines under Mitsunobu reaction conditions. This cross-coupling reaction, in which only two steps needed using 3,4-dihydropyrimidinones as starting material, was superior to other reported processes.

b Isolated yields.

$$RO_{2}C$$

$$RO_{$$

Scheme 3. A proposed mechanism for the formation of C2-multifunctionalized pyrimidines.

3. Conclusion

In conclusion, we have developed a novel and efficient synthetic method to prepare *C2*-multifunctionalized pyrimidines by the Mitsunobu reaction between 2-hydroxy pyrimidine and alcohols, amines, phenols and carboxylic acids nucleophiles. Compared to previously known approaches, the simplicity (only two-step needed using 3,4-dihydropyrimidinones as starting material) and higher efficiency make this method particularly attractive. These results provided, as well as other reported studies, that Mitsunobu reaction conditions could be potentially applicable to other electron-deficient heterocyclic or aromatic systems. The present study also provides a readily accessible approach to construct multifunctionalized pyrimidine template for diversity-oriented synthesis.

4. Experimental section

4.1. General information

Commercially available reagents were used without further purification. Solvents were treated prior to use according to the standard methods. Column chromatography was performed on silica gel (300–400 mesh). Melting points were determined on an XT-4 electrothermal micromelting point apparatus and are uncorrected. NMR spectra were recorded at 400 (¹H) and 100 (¹³C) MHz, respectively, on a Varian Mercury plus-400 instrument using CDCl₃ as solvent and TMS as an internal standard. Mass spectra were obtained on a Bruker Daltonics APEXII 47e FT-ICR spectrometer. The Biginelli DHPM 1 and their dehydrogenated compounds 2 were readily prepared according to the procedures by Fu et al. ¹¹ and Yamamoto et al. ⁹

4.2. General procedure for the synthesis of ${\it C2}$ -functionalized pyrimidines

At 0 $^{\circ}$ C, DIAD (1.5 mmol) was dissolved in THF (3 mL), followed by the addition of TPP (1.5 mmol) with the forming of a suspension, then 2-hydroxy pyrimidine (1.0 mmol) and nucle-ophile (1.5 mmol) were added. The mixture was warmed to rt and stirred for further 2–6 h. After completion of the reaction monitored by thin layer chromatography (TLC), the solvent was evaporated under reduced pressure, and then the crude product was purified by column chromatography over silica gel with ethyl acetate and petroleum ether as the eluent, to give target product **4, 6** or **8**.

4.2.1. Ethyl 4-methyl-6-phenyl-2-(piperidin-1-yl)pyrimidine-5-carboxylate (**4a**). Obtained 273 mg, yield 84%, white solid, mp 69–70 °C. 1 H NMR (400 MHz, CDCl₃) δ =7.37–7.58 (m, 5H), 4.02 (q, J=6.8 Hz, 2H), 3.89 (t, J=5.6 Hz, 4H), 2.49 (s, 3H), 1.62 (m, 6H), 0.93 (t, J=7.2 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =169.2, 166.9, 165.7,

160.1, 139.7, 129.2, 128.0, 113.1, 60.7, 44.6, 25.8, 24.8, 23.2, 13.5. ESI-MS; m/z 326 ([M+H $^+$]).

4.2.2. Ethyl 4-(4-methoxyphenyl)-6-methyl-2-(piperidin-1-yl)pyrimidine-5-carboxylate (**4b**). Obtained 287 mg, yield 81%, white solid, mp 87–88 °C. 1 H NMR (400 MHz, CDCl₃) δ =7.56 (d, J=8.8 Hz, 2H), 6.91 (d, J=8.8 Hz, 2H), 4.11 (q, J=7.6 Hz, 2H), 3.88 (t, J=5.2 Hz, 4H), 3.84 (s, 3H), 2.47 (s, 3H), 1.60–1.68 (m, 6H), 1.03 (t, J=7.2 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =169.5, 166.5, 164.7, 160.7, 132.0, 129.7, 113.5, 60.8, 55.3, 44.6, 25.8, 24.8, 23.1, 13.7. ESI-MS: m/z 356 (JM+H⁺J).

4.2.3. Ethyl 4-methyl-2-(piperidin-1-yl)-6-p-tolylpyrimidine-5-carboxylate (**4c**). Obtained 281 mg, yield 83%, white solid, mp 68–69 °C. ¹H NMR (400 MHz, CDCl₃) δ =7.47 (d, J=8.0 Hz, 2H), 7.19 (d, J=8.0 Hz, 2H), 4.08 (q, J=6.8 Hz, 2H), 3.88 (t, J=5.2 Hz, 4H), 2.47 (s, 3H), 2.38 (s, 3H), 1.59–1.67 (m, 6H), 0.99 (t, J=7.2 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ =169.4, 166.6, 165.5, 139.3, 136.8, 128.7, 128.0, 60.7, 44.6, 25.8, 24.8, 23.1, 21.3, 13.6. ESI-MS: m/z 340 ([M+H⁺]).

4.2.4. Ethyl 4-(4-chlorophenyl)-6-methyl-2-(piperidin-1-yl)pyrimidine-5-carboxylate (4d). Obtained 280 mg, yield 78%, white solid, mp 54–55 °C. 1 H NMR (400 MHz, CDCl₃) δ =7.50 (d, J=8.4 Hz, 2H), 7.36 (d, J=8.4 Hz, 2H), 4.07 (m, J=6.8 Hz, 2H), 3.88 (t, J=5.2 Hz, 4H), 2.48 (s, 3H), 1.60–1.68 (m, 6H), 1.01 (t, J=7.2 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =169.4, 166.6, 165.5, 139.3, 136.8, 128.7, 128.0, 113.1, 60.7, 44.6, 25.8, 24.8, 23.1. ESI-MS: m/z 360 ([M+H+]).

4.2.5. Ethyl 4-(4-bromophenyl)-6-methyl-2-(piperidin-1-yl)pyrimidine-5-carboxylate (4e). Obtained 342 mg, yield 85%, white solid, mp 61–62 °C. 1 H NMR (400 MHz, CDCl₃) δ =7.52 (d, J=8.4 Hz, 2H), 7.43 (d, J=8.0 Hz, 2H), 4.05 (q, J=7.2 Hz, 2H), 3.88 (t, J=5.2 Hz, 4H), 2.49 (s, 3H), 1.60–1.68 (m, 6H), 1.00 (t, J=7.2 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =168.8, 167.2, 164.5, 160.0, 138.7, 131.2, 129.7, 123.6, 60.8, 44.6, 25.8, 24.8, 23.2, 13.6. ESI-MS: m/z 404 ([M+H⁺]).

4.2.6. Ethyl 4-(4-fluorophenyl)-6-methyl-2-(piperidin-1-yl)pyrimidine-5-carboxylate (**4f**). Obtained 295 mg, yield 86%, white solid, mp 51–52 °C. ¹H NMR (400 MHz, CDCl₃) δ =7.06–7.59 (m, 4H), 4.06 (q, J=6.8 Hz, 2H), 3.88 (t, J=5.6 Hz, 4H), 2.48 (s, 3H), 1.61 (m, 6H), 1.00 (t, J=6.8 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ =169.3, 167.3, 165.0, 164.7, 162.5, 160.3, 136.0, 130.4, 130.2, 115.4, 115.1, 113.2, 61.1, 44.9, 26.1, 25.0, 23.4, 13.8. ESI-MS: m/z 344 ([M+H⁺]).

4.2.7. Methyl 4-(4-fluorophenyl)-6-isopropyl-2-(piperidin-1-yl)pyrimidine-5-carboxylate (**4g**). Obtained 286 mg, yield 80%, white solid, mp 96–97 °C. 1 H NMR (400 MHz, CDCl₃) δ =7.06–7.60 (m, 4H), 3.89 (t, J=5.2 Hz, 4H), 3.59 (s, 3H), 3.19 (m, 1H), 1.58–1.68 (m, 6H), 1.24 (d, J=6.4 Hz, 6H). 13 C NMR (100 MHz, CDCl₃) δ =174.4, 170.0, 164.7, 163.8, 162.2, 160.5, 135.7, 129.9, 115.2, 115.0, 112.4, 51.9, 44.7, 32.7, 25.7, 24.8, 21.7. ESI-MS: m/z 358 ([M+H $^+$]).

4.2.8. Ethyl 4-methyl-2-morpholino-6-phenylpyrimidine-5-carboxyl-ate (4h). Obtained 268 mg, yield 82%, white solid, mp 118–119 °C. ¹H

- NMR (400 MHz, CDCl₃) δ =7.40–7.58 (m, 5H), 4.04 (q, J=7.2 Hz, 2H), 3.76 (t, J=4.8 Hz, 4H), 2.50 (s, 3H), 0.95 (t, J=7.6 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ =168.9, 167.0, 165.7, 160.2, 139.3, 129.5, 128.2, 128.1, 114.5, 66.9, 60.9, 44.1, 23.1, 13.5. ESI-MS: m/z 328 ([M+H $^+$]).
- 4.2.9. Ethyl 4-(4-methoxyphenyl)-6-methyl-2-morpholinopyrimidine-5-carboxylate (**4i**). Obtained 286 mg, yield 80%, white solid, mp 72–74 °C. 1 H NMR (400 MHz, CDCl₃) δ =7.58(d, J=9.2 Hz, 2H), 6.93 (d, J=8.4 Hz, 2H), 4.13 (q, J=6.8 Hz, 2H), 3.84 (s, J=4.8 Hz, 3H), 3.75 (t, J=4.8 Hz, 4H), 2.47 (s, 3H), 1.05 (t, J=7.2 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =169.3, 166.6, 164.7, 160.9, 160.2, 131.5, 129.7, 114.2, 113.6, 66.9, 61.0, 55.3, 55.3, 44.1, 23.1, 13.7. ESI-MS: m/z 358 ([M+H⁺]).
- 4.2.10. Ethyl 4-methyl-2-morpholino-6-p-tolylpyrimidine-5-carboxylate (4j). Obtained 273 mg, yield 80%, white solid, mp 68–70 °C. 1 H NMR (400 MHz, CDCl₃) δ =7.49 (d, J=8.0 Hz, 2H), 7.22 (d, J=8.0 Hz, 2H), 4.10 (q, J=7.2 Hz, 2H), 3.92 (t, J=4.4 Hz, 4H), 3.75 (t, J=4.4 Hz, 4H), 2.48 (s, 4H), 2.38 (s, 3H), 1.01 (t, J=6.8 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =169.2, 166.7, 165.4, 160.2, 139.6, 136.4, 128.9, 128.1, 128.0, 114.5, 66.9, 60.9, 44.1, 23.0, 21.3, 13.6. ESI-MS: m/z 342 ([M+H⁺]).
- 4.2.11. Ethyl 4-(4-chlorophenyl)-6-methyl-2-morpholinopyrimidine-5-carboxylate (**4k**). Obtained 299 mg, yield 83%, white solid, mp 80–83 °C. 1 H NMR (400 MHz, CDCl₃) δ =7.37–7.52 (m, 4H), 4.10 (m, J=7.2 Hz, 2H), 3.92 (t, J=5.6 Hz, 4H), 3.76 (t, J=5.2 Hz, 4H), 2.50 (s, 3H), 1.02 (t, J=7.6 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =168.7, 167.2, 164.5, 160.0, 137.7, 135.7, 129.5, 128.4, 114.3, 66.8, 61.1, 44.1, 23.2, 13.7. ESI-MS: m/z 362 ([M+H⁺]).
- 4.2.12. Ethyl 4-(4-bromophenyl)-6-methyl-2-morpholinopyrimidine-5-carboxylate (41). Obtained 324 mg, yield 80%, white solid, mp 92–93 °C. ¹H NMR (400 MHz, CDCl₃) δ =7.43–7.55 (m, 4H), 4.08 (q, J=7.2 Hz, 2H), 3.91 (t, J=4.4 Hz, 4H), 3.75 (t, J=4.8 Hz, 4H), 2.49 (s, 3H), 1.02 (t, J=6.8 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ =168.7, 167.3, 164.5, 160.1, 138.3, 131.4, 129.8, 123.9, 114.3, 66.8, 61.1, 44.2, 23.2, 13.7. ESI-MS: m/z 406 ([M+H⁺]).
- 4.2.13. Ethyl 4-(4-fluorophenyl)-6-methyl-2-morpholinopyrimidine-5-carboxylate (4m). Obtained 283 mg, yield 82%, white solid, mp 86–89 °C. 1 H NMR (400 MHz, CDCl₃) δ =7.07–7.59 (m, 4H), 4.10 (q, J=7.2 Hz, 2H), 3.92 (t, J=4.8 Hz, 4H), 3.76 (t, J=4.8 Hz, 4H), 2.49 (s, 3H), 1.02 (t, J=7.2 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =168.8, 167.1, 164.9, 164.4, 162.4, 160.1, 135.3, 130.1, 130.1, 115.3, 115.1, 114.3, 66.8, 61.1, 44.1, 23.2, 13.7. ESI-MS: m/z 346 ([M+H⁺]).
- 4.2.14. Methyl 4-(4-fluorophenyl)-6-isopropyl-2-morpholinopyrimidine-5-carboxylate (**4n**). Obtained 301 mg, yield 84%, white solid, mp 102–103 °C. ^1H NMR (400 MHz, CDCl₃) δ =7.08–7.61 (m, 4H), 3.92 (t, J=4.4 Hz, 4H), 3.77 (t, J=5.2 Hz, 4H), 3.61 (s, 3H), 3.19 (m, 1H), 1.26 (d, J=6.4 Hz, 6H). ^{13}C NMR (100 MHz, CDCl₃) δ =174.6, 169.7, 164.9, 163.9, 162.4, 160.6, 135.3, 130.0, 115.4, 115.2, 113.7, 66.8, 52.1, 44.2, 32.9, 21.7. ESI-MS: m/z 360 ([M+H $^+$]).
- 4.2.15. Ethyl 2-(ethylamino)-4-methyl-6-phenylpyrimidine-5-carboxylate (**40**). Obtained 219 mg, yield 77%, white solid, mp 70–72 °C. 1 H NMR (400 MHz, CDCl₃) δ =7.38–7.54 (m, 5H), 5.39 (s, 1H), 4.05 (q, J=6.8 Hz, 2H), 3.51 (m, 2H), 2.49 (s, 3H), 1.22 (t, J=7.6 Hz, 3H), 0.94 (t, J=7.2 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =168.8, 167.1, 166.1, 161.1, 139.2, 129.3, 128.1, 127.9, 115.0, 60.9, 36.2, 22.9, 14.8, 13.5. ESI-MS: m/z 286 ([M+H $^+$]).
- 4.2.16. Ethyl 2-ethoxy-4-methyl-6-phenylpyrimidine-5-carboxylate (**6a**). Obtained 271 mg, yield 95%, colorless oil. 1 H NMR (400 MHz, CDCl₃) δ =7.30–7.56 (m, 5H), 4.40 (m, 2H), 4.05 (m, 2H), 2.47 (s, 3H), 1.34 (t, J=6.8 Hz, 3H), 0.92 (t, J=7.2 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =168.4, 168.0, 166.2, 163.8, 137.6, 129.7,

- 128.1, 128.0, 119.4, 63.4, 61.3, 22.5, 14.2, 13.3. ESI-MS: m/z 287 ([M+H⁺]).
- 4.2.17. Ethyl 2-isopropoxy-4-methyl-6-phenylpyrimidine-5-carboxylate (**6b**). Obtained 279 mg, yield 93%, colorless oil. 1 H NMR (400 MHz, CDCl₃) δ =7.38–7.65 (m, 5H), 5.41 (m, 1H), 4.14 (m, 2H), 2.55 (s, 3H), 1.38 (m, 6H), 1.06 (t, J=6.8 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =168.4, 168.1, 166.2, 163.5, 137.7, 129.7, 128.1, 128.0, 119.2, 61.3, 22.6, 22.5, 21.6, 21.3, 13.3. ESI-MS: m/z 301 ([M+H⁺]).
- 4.2.18. Ethyl 2-(benzyloxy)-4-methyl-6-phenylpyrimidine-5-carboxylate (**6c**). Obtained 313 mg, yield 90%, colorless oil. 1 H NMR (400 MHz, CDCl₃) δ =7.30–7.63 (m, 10H), 5.51 (s, 2H), 4.15 (q, J=7.2 Hz, 2H), 2.58 (s, 3H), 1.02 (t, J=7.2 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =168.6, 168.1, 166.3, 163.7, 137.6, 136.2, 129.9, 128.4, 128.2, 128.2, 128.1, 127.9, 119.9, 69.2, 61.5, 22.6, 13.4. ESI-MS: m/z 349 ([M+H $^{+}$]).
- 4.2.19. Ethyl 4-(4-chlorophenyl)-2-ethoxy-6-methylpyrimidine-5-carboxylate (6d). Obtained 307 mg, yield 96%, pale yellow oil. $^{1}\mathrm{H}$ NMR (400 MHz, CDCl₃) δ =7.69 (dd, J=2 Hz, 6.4 Hz, 2H), 7.41 (dd, J=2 Hz, 6.0 Hz, 2H), 4.50 (q, J=6.8 Hz, 2H), 4.5018 (q, J=6.8 Hz, 2H), 2.57 (s, 3H), 1.45 (t, J=6.8 Hz, 3H), 1.10 (t, J=6.8 Hz, 3H). $^{13}\mathrm{C}$ NMR (100 MHz, CDCl₃) δ =168.8, 168.0, 165.2, 164.0, 136.3, 136.3, 129.6, 128.5, 119.5, 63.7, 61.6, 22.7, 14.3, 13.6. ESI-MS: m/z 320 M $^{+}$, 322 [M $^{+}\mathrm{C}$] $^{+}$. Anal. Calcd for $C_{16}\mathrm{H}_{17}\mathrm{ClN}_{2}\mathrm{O}_{3}$: C, 59.91; H, 5.34; N, 8.73. Found: C, 59.69; H, 5.31; N, 8.70.
- 4.2.20. Ethyl 2-(4-nitrophenoxy)-4-methyl-6-phenylpyrimidine-5-carboxylate (6e). Obtained 310 mg, yield 82%, white solid, mp 118–119 °C. ¹H NMR (400 MHz, CDCl₃) δ =8.33–8.29 (m, 2H), 7.60–7.50 (m, 2H), 7.49–7.27 (m, 5H), 4.23–4.18 (m, 2H), 2.59 (s, 3H), 1.09–1.06 (m, 3H). ¹³C NMR (100 MHz, CDCl₃) δ =169.6, 167.6, 166.8, 162.9, 157.6, 144.8, 136.8, 130.6, 128.6, 128.3, 125.3, 122.1, 62.0, 22.7, 13.6. ESI-MS: m/z 379 M⁺. Anal. Calcd for C₂₀H₁₇N₃O₅: C, 63.32; H, 4.52; N, 11.08. Found: C, 63.02; H, 4.50; N, 11.02.
- 4.2.21. Ethyl 2-(4-nitrophenoxy)-4-(4-methoxyphenyl)-6-methyl-pyrimidine-5-carboxylate (**6f**). Obtained 388 mg, yield 95%, white solid, mp 105–107 °C. 1 H NMR (400 MHz, CDCl₃) δ =8.30 (dd, J=2 Hz, 6.8 Hz, 2H), 7.59 (dd, J=2 Hz, 6.8 Hz, 2H), 7.41 (dd, J=2 Hz, 6.8 Hz, 2H), 6.93 (dd, J=2 Hz, 6.8 Hz, 2H), 4.27 (q, J=7.2 Hz, 2H), 3.85 (s, 3H), 2.57 (s, 3H), 1.17 (t, J=7.2 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =169.2, 168.1, 165.8, 162.8, 161.8, 157.7, 144.7, 130.1, 128.9, 125.3, 122.1, 121.4, 114.0, 62.0, 55.4, 22.6, 13.8. ESI-MS: m/z 409 M⁺. Anal. Calcd for C₂₁H₁₉N₃O₆: C, 61.61; H, 4.68; N, 10.26. Found: C, 61.29; H, 4.65; N, 10.20.
- 4.2.22. Ethyl 2-(4-nitrophenoxy)-4-(4-chlorophenyl)-6-methylpyrimidine-5-carboxylate ($\mathbf{6g}$). Obtained 355 mg, yield 86%, white solid, mp 121–123 °C. 1 H NMR (400 MHz, CDCl₃) δ =8.31 (dd, J=2 Hz, 6.8 Hz, 2H), 7.54 (dd, J=1.6 Hz, 6.8 Hz, 2H), 7.43–7.39 (m, 4H), 4.23 (q, J=7.2 Hz, 2H), 2.59 (s, 3H), 1.14 (t, J=7.2 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =169.8, 167.4, 165.5, 162.9, 157.5, 144.9, 137.1, 135.2, 129.7, 128.9, 125.4, 122.1, 121.9, 62.2, 22.8, 13.7. ESI-MS: m/z 413 M $^{+}$, 415 [M+2] $^{+}$. Anal. Calcd for C $_{20}$ H $_{16}$ ClN $_{3}$ O $_{5}$: C, 58.05; H, 3.90; N, 10.15. Found: C, 58.28; H, 3.93; N, 10.20.
- 4.2.23. Ethyl 2-(4-nitrophenoxy)-4-(4-fluorophenyl)-6-isopropylpyrimidine-5-carboxylate (*6h*). Obtained 329 mg, yield 80%, colourless oil.

 ¹H NMR (400 MHz, CDCl₃) δ =8.34–8.30 (m, 2H), 7.65–7.60 (m, 2H), 7.45–7.42 (m, 2H), 7.28–7.10 (m, 2H), 3.74 (s, 3H), 3.22–3.15 (m, 1H), 1.32–1.14 (m, 6H).

 ¹³C NMR (100 MHz, CDCl₃) δ =177.5, 168.3, 165.5, 163.4, 162.9, 157.6, 144.8, 132.9, 130.4, 125.2, 122.1, 120.9, 115.9, 115.7, 52.9, 33.5, 21.6. ESI-MS: m/z 411 M⁺. Anal. Calcd for C₂₂H₂₀FN₃O₅: C, 62.11; H, 4.74; N, 9.88, Found: C, 62.36; H, 4.72; N, 9.83.
- 4.2.24. Ethyl 2-(4-chlorophenoxy)-4-(4-chlorophenyl)-6-methylpyrimidine-5-carboxylate (6i). Obtained 313 mg, yield 78%, white solid,

mp 97–99 °C. ¹H NMR (400 MHz, CDCl₃) δ =7.55–7.52 (m, 2H), 7.41–7.35 (m, 4H), 7.18–7.16 (m, 2H), 4.24–4.19 (m, 2H), 2.56 (s, 3H), 1.15–1.11 (m, 3H). ¹³C NMR (100 MHz, CDCl₃) δ =169.5, 167.7, 165.3, 163.7, 151.2, 136.8, 135.5, 130.6, 129.7, 129.5, 128.8, 122.9, 121.1, 62.0, 22.8, 13.7. ESI-MS: m/z 402 M⁺, 404 [M+2]⁺. Anal. Calcd for C₂₀H₁₆Cl₂N₂O₃: C, 59.57; H, 4.00; N, 6.95. Found: C, 59.82; H, 4.02; N, 6.98.

4.2.25. Ethyl 2-(benzoyloxy)-4-methyl-6-phenylpyrimidine-5-carboxylate (**8a**). Obtained 254 mg, yield 70%, white solid, mp 89–90 °C. 1 H NMR (400 MHz, CDCl₃) δ =8.25–8.23 (m, 2H), 7.69–7.63 (m, 3H), 7.51–7.43 (m, 5H), 4.23 (q, J=7.2 Hz, 2H), 2.67 (s, 3H), 1.09 (t, J=7.2 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =169.3, 167.4, 167.1, 164.1, 160.3, 136.7, 134.1, 130.6, 130.5, 128.6, 128.5, 128.4, 124.3, 62.1, 22.6, 13.6. ESI-MS: m/z 363 M $^{+}$. Anal. Calcd for C₂₁H₁₈N₂O₄: C, 69.60; H, 5.01; N, 7.73. Found: C, 69.86; H, 5.00; N, 7.76.

4.2.26. Ethyl 2-(benzoyloxy)-4-(4-methoxyphenyl)-6-methylpyrimidine-5-carboxylate ($\bf 8b$). Obtained 267 mg, yield 68%, colourless oil. 1 H NMR (400 MHz, CDCl₃) δ =8.25–8.23 (m, 2H), 7.72–7.66 (m, 2H), 7.64–7.62 (m, 1H), 7.52–7.49 (m, 2H), 6.98–6.95 (m, 2H), 4.28 (q, $\it J$ =6.8 Hz, 2H), 3.84 (s, 3H), 2.65 (s, 3H), 1.18 (t, $\it J$ =6.8 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =168.9, 167.8, 166.1, 164.1, 161.7, 160.2, 134.0, 130.5, 130.2, 128.9, 128.6, 128.5, 123.6, 114.0, 62.0, 55.3, 22.5, 13.7. ESI-MS: $\it m/z$ 392 M $^+$. Anal. Calcd for C₂₂H₂₀N₂O₅: C, 67.34; H, 5.14; N, 7.14. Found: C, 67.04; H, 5.13; N, 7.17.

4.2.27. Ethyl 2-(benzoyloxy)-4-(4-chlorophenyl)-6-methylpyrimidine-5-carboxylate (**8c**). Obtained 273 mg, yield 69%, colourless oil. 1 H NMR (400 MHz, CDCl₃) δ =8.25-8.22 (m, 2H), 7.67-7.63 (m, 3H), 7.53-7.49 (m, 2H), 7.45-7.43 (m, 2H), 4.26 (q, J=6.8 Hz, 2H), 2.67 (s, 3H), 1.16 (t, J=6.8 Hz, 3H). 13 C NMR (100 MHz, CDCl₃) δ =169.6, 167.2, 165.6, 164.1, 160.3, 137.0, 135.1, 130.6, 129.9, 128.9, 128.6, 128.5, 124.1, 62.3, 22.6, 13.7. ESI-MS: m/z 396 M⁺, 398 [M+2]⁺. Anal. Calcd for C₂₁H₁₇ClN₂O₄: C, 63.56; H, 4.32; N, 7.06. Found: C, 63.86; H, 4.34; N, 7.08.

4.2.28. Ethyl 2-(benzoyloxy)-4-(4-fluorophenyl)-6-isopropylpyrimidine-5-carboxylate (**8d**). Obtained 248 mg, yield 63%, colourless oil. 1 H NMR (400 MHz, CDCl₃) δ =8.26–8.24 (m, 2H), 7.73–7.69 (m, 2H), 7.67–7.63 (m, 1H), 7.54–7.49 (m, 2H), 7.17–7.12 (m, 2H), 3.78 (s, 3H), 3.20 (t, J=7.2 Hz, 2H), 1.36 (s, 3H), 1.34 (s, 3H). 13 C NMR (100 MHz, CDCl₃) δ =177.4, 168.2, 165.5, 165.4, 164.0, 162.9, 160.9, 134.1, 132.9, 130.6, 130.5, 128.6, 128.5, 123.0, 115.9, 115.7, 52.9, 33.6, 21.6. ESI-MS: m/z 394 M⁺, 396 [M+2]⁺. Anal. Calcd for C_{23} H₂₁FN₂O₄: C, 67.64; H, 5.18; N, 6.86. Found: C, 67.89; H, 5.21; N, 6.89.

4.2.29. Ethyl 2-(4-methoxybenzoyloxy)-4-(4-chlorophenyl)-6-methylpyrimidine-5-carboxylate (**8e**). Obtained 307 mg, yield 72%, white solid, mp 99–100 °C. 1 H NMR (400 MHz, CDCl₃) δ =8.18 (dd, J=2 Hz, 6.8 Hz, 2H), 7.64 (dd, J=2 Hz, 4.8 Hz, 2H), 7.43 (dd, J=2 Hz, 4.8 Hz, 2H), 6.98 (dd, J=1.6 Hz, 7.2 Hz, 2H), 4.28–4.23 (m, 2H), 3.90 (s, 3H), 2.66 (s, 3H), 1.18–1.14 (m, 3H). 13 C NMR (100 MHz, CDCl₃) δ =169.5, 167.3, 165.6, 164.4, 160.5, 136.9, 135.1, 132.9, 129.9, 128.9, 123.9, 120.6, 113.9, 62.2, 55.5, 22.7, 13.7. ESI-MS: m/z 426 M⁺, 428 [M+2]⁺. Anal. Calcd for C₂₂H₁₉ClN₂O₅: C, 61.90; H, 4.49; N, 6.56. Found: C, 61.61; H, 4.45; N, 6.51.

4.2.30. Ethyl 2-(4-nitrobenzoyloxy)-4-(4-chlorophenyl)-6-methyl-pyrimidine-5-carboxylate (**8f**). Obtained 353 mg, yield 80%, pale yellow oil. ¹H NMR (400 MHz, CDCl₃) δ =8.43–8.36 (m, 4H), 7.65 (dd, J=2 Hz, 6.8 Hz, 2H), 7.45 (dd, J=2 Hz, 6.4 Hz, 2H), 4.28 (q, J=6.8 Hz, 2H), 2.69 (s, 3H), 1.17 (t, J=6.8 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃) δ =169.9, 166.9, 165.8, 162.2, 159.7, 151.2, 137.3, 134.8, 133.9, 131.7, 129.8, 129.0, 124.6, 123.8, 62.4, 22.7, 13.7.

ESI-MS: m/z 441 M⁺, 443 [M+2]⁺. Anal. Calcd for $C_{21}H_{16}ClN_3O_6$: C, 57.09; H, 3.65; N, 9.51. Found: C, 57.30; H, 3.67; N, 9.56.

4.2.31. Ethyl 2-(4-chlorobenzoyloxy)-4-(4-chlorophenyl)-6-methylpyrimidine-5-carboxylate (**8g**). Obtained 339 mg, yield 79%, yellow oil. 1 H NMR (400 MHz, CDCl₃) δ =8.16 (dd, J=2 Hz, 4.8 Hz, 2H), 7.63 (dd, J=1.6 Hz, 5.6 Hz, 2H), 7.49 (dd, J=2.4 Hz, 4.8 Hz, 2H), 7.44 (dd, J=2.0 Hz, 4.8 Hz, 2H), 4.29–4.21 (m, 2H), 2.67 (s, 3H), 1.18–1.08 (m, 3H). 13 C NMR (100 MHz, CDCl₃) δ =169.7, 167.1, 165.7, 163.2, 160.0, 140.8, 137.1, 134.9, 131.9, 129.8, 129.6, 129.0, 128.9, 128.6, 126.9, 124.3, 62.3, 22.6, 13.7. ESI-MS: m/z 430 M+, 432 [M+2]+. Anal. Calcd for C21H16Cl2N2O4: C, 58.48; H, 3.74; N, 6.50. Found: C, 58.20; H, 3.71; N, 6.53.

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Supplementary data

Supplementary data related to this article can be found online at doi:10.1016/j.tet.2011.02.046. These data include MOL files and InChIKeys of the most important compounds described in this article.

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